

**KREMNIY Si(100)2×1 SIRTIDA ADSORBSIYALANGAN C₂₀@C_N VA C₆₀@C_N (N=1, 3)
EKZOFULLERENLARINI FULLERENLAR ADSORBSIYASI MAVJUD BO'LMAGANDA
VAQT PARAMETRLARI BO'YICHA BAHOLASH**

I.Z. Urolov², A.A. Dhzurakhalov², I.D. Yadgarov¹

¹*Institute of Ion-Plasma and Laser Technologies named after U.A. Arifov, Uzbekistan Academy
Sciences, Tashkent, 100125 Uzbekistan.*

²*University of Antwerp, Prinsstraat 13, 2000 Antwerp, Belgium*

e-mail: ishmuminyadgarov@gmail.com

Annotatsiya. Ushbu tadqiqot ishida Si(100)2×1 sirtining dimer qatorida C_n (n=1, 3) uglerod klasterlari orqali adsorbsiyalangan C₂₀@C_n va C₆₀@C_n ekzofullerenlarining fullerenlar adsorbsiyasi mavjud bo'lmaganda C₂₀ va C₆₀ fulleren molekulalarining C_n uglerod klasterlarini fulleren molekulasiga ekzodrillovchi yakka C-C bog' bo'ylab o'tuchi o'q atrofidagi aylanma va tebranma (presetsion) harakat chastotalarini past temperaturalar sohasida ($T < 100$ K) temperaturaga bog'lanishi molekulyar dinamika usuli asosida ochiq manba simulyatori LAMMPS paket dasturida modellashtirilgan holda o'rganildi.

Аннотация. В данной исследовательской работе изучено температурное ($T < 100$ K) поведение частот вращательных и колебательных (прецессионных) движений вокруг оси, проходящей вдоль одиночной экзодриллирующей C–C связи углеродных кластеров C_n ($n = 1, 3$), адсорбированных на димерном ряду поверхности Si(100)2×1 в составе экзофуллеренов C₂₀@C_n и C₆₀@C_n, а также в отсутствие адсорбции фуллеренов. Исследование выполнено методом молекулярной динамики с использованием открытого симуляционного пакета LAMMPS.

Abstract. In this research work, the temperature dependence ($T < 100$ K) of the rotational and oscillatory (precessional) motion frequencies around the axis passing through the single exo-drilling C–C bond of carbon clusters C_n ($n = 1, 3$), adsorbed on the dimer row of the Si(100)2×1 surface as part of the exofullerenes C₂₀@C_n and C₆₀@C_n, as well as in the absence of fullerene adsorption, was studied. The study was carried out using the molecular dynamics method and modeled in the open-source simulator LAMMPS.

Kalit so'zlar: fulleren, kremniy, adsorbsiya, aylanma harakat, presetsiya, chastota, temperatura, energiya, bog', modellashtirish.

Ключевые слова: фуллерен, кремний, адсорбция, вращательное движение, прецессия, частота, температура, энергия, связь, моделирование.

Keywords: fullerene, silicon, adsorption, rotational motion, precision, frequency, temperature, energy, bond, modelling.

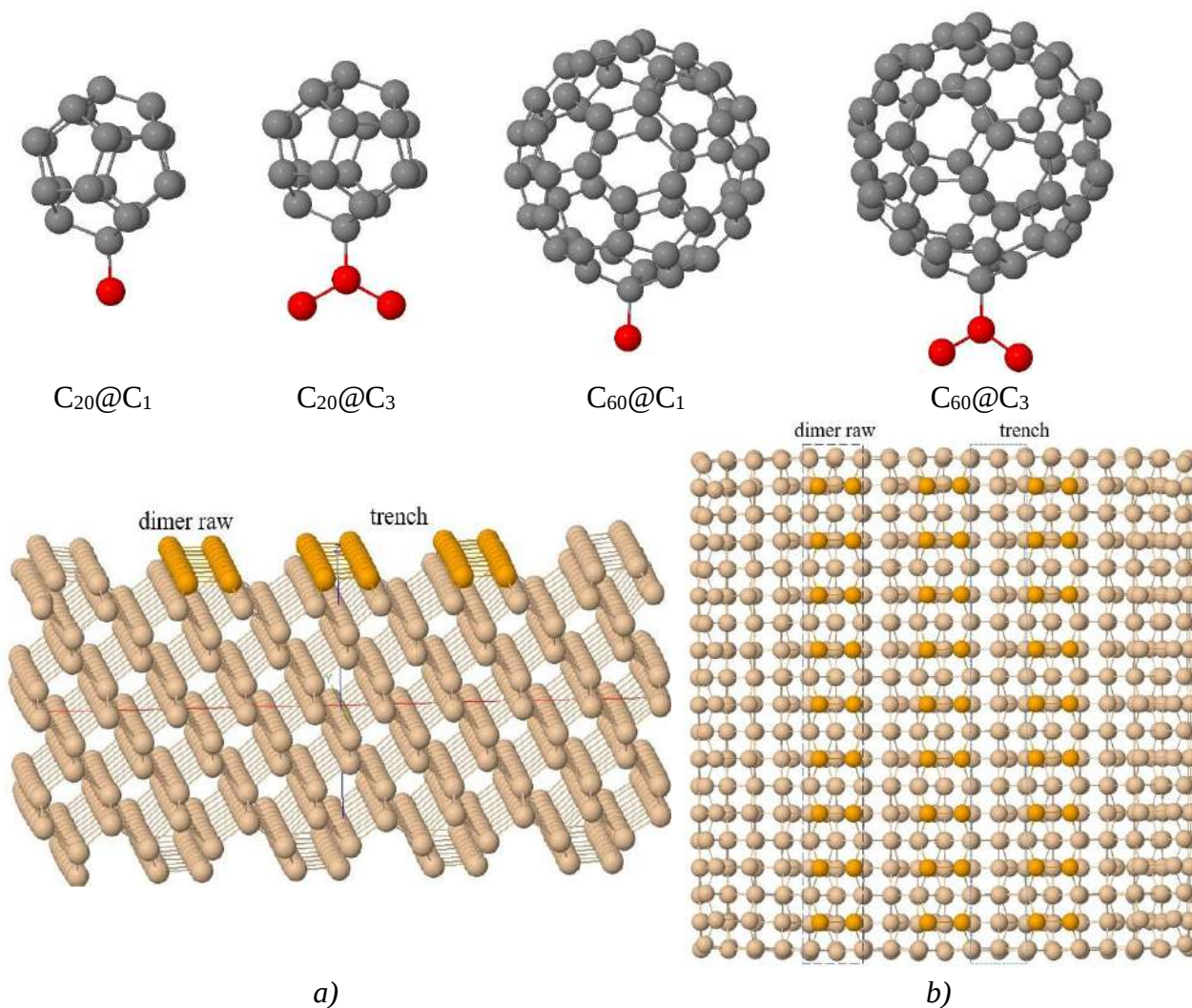
Fulleren molekulalari adsorbsiyalanishidan yuzaga keluvchi nanostrukturalar o'zining noyob fizik va kimyoviy xususiyatlari sababli bugungi kun ilmiy izlanishlarning muhim tadqiqot sohalaridan biri sifatida ilmiy izlanuvchilar e'tiborini jalb qilib kelayotgan yo'nalishlardan biri hisoblanadi. Jumladan, $(100)2\times 1$ rekonstruksiya kremniy taglik sirtida C_{20} va C_{60} molekulalarining C_n uglerod klasterlari yordamida adsorbsiyalanishidan yuzaga keluvchi gibrid materiallar yarimo'tkazgichli qurilmalar o'rnini bosuvchi molekulyar qurilmalar ishlab chiqarishda ahamiyatli hisoblansa [1], organik quyosh elementlarining muhim tarkibiy qismi bo'lgan fullerene asosli gibrid materiallarda fullerene C_{60} molekulasining qatlam sirtida adsorbsiyalanishining mustahkam bo'lishligi katta ahamiyat kasb etadi [2] va $C_{20}@C_n$ va $C_{60}@C_n$ ($n=1-5$) ekzofullerenlari C_{20} va C_{60} fullerene molekulalariga nisbatan kremniy Si(100) taglik sirtida qattiq yoki kovalent bog'lanishli Si-C bog'lar hosil qilgan holda adsorbsiyalanar ekan [3]. Mazkur omillarning mavjudligi o'laroq, kremniy tagliklar sirtida C_n uglerod klasterlari yordamida adsorbsiyalangan $C_{20}@C_n$ va $C_{60}@C_n$ ekzofullerenlarini fullerene adsorbsiyasi mavjud bo'lmaganda vaqt parametrlari bo'yicha baholash orqali tadqiq etish bugungi kun fullerene asosidagi qurilmalar (molekulyar tranzistorlar, sensorlar, kvant komponentlar) uchun molekulalarning sirtga nisbatan yo'nalishi va dinamik barqarorligi funksional xususiyatlarni belgilaydi.

C_{20} va C_{60} fullerene molekulalarining C-C bog' bo'ylab o'tuvchi o'q atrofidagi aylanma va ma'lum yo'nalishdagi soda tebranma (presetsion) harakatlariga o'xshash harakatlar ochiq zanjirli alkanlar (C_nH_{2n+2}) guruhining barchasida fullerene molekulalaridagi singari σ bog'lanishli markaziy C-C bog' atrofida kuzatiladi [4-9]. Alkanlarda CH_3 guruhlarining ichki aylanma harakatiga potensial to'siqlar mavjud, masalan, etan uchun ushbu qiymat ≈ 12 kJ/mol ni tashkil qiladi va u maksimal va minimal konformatsiyalar orasidagi energiya farqlari hisobidan yuzaga kelib, ushbu potensial energiya burilish (dihedral) burchagining funksiyasi bo'lib, uto'liq 360° aylanish doirasida uchta ekvivalent minima (qadamli) va uchta maksima (tutilgan) ni ko'rsatadi [5]. Steric/repulsion modeliga ko'ra, eclipsed holatda bitta CH_3 dagi vodorodlar ikkinchi CH_3 dagi vodorodlarga yaqinlashadi, bu esa H-H itarilishga olib kelgan holda ta'sirlashuv energiyasini oshiradi [10].

Mazkur tadqiqot ishida Si(100) 2×1 sirtning dimer qatorlarida C_n ($n=1, 3$) uglerod klasterlari orqali adsorbsiyalangan $C_{20}@C_n$ va $C_{60}@C_n$ ekzofullerenlarining C_{20} va C_{60} fullerene molekulalarining bevosita Si(100) 2×1 sirtida adsorbsiyasi kuzatilmagan hollarda ushbu fullerene larning C-C bog' bo'ylab o'tuvchi o'q atrofidagi aylanma va shu o'qga nisbatan tebranma harakatlari molekulyar dinamika (MD) usulida modellashtirish orqali bu ikki harakatning

chastotalarini past temperaturalar sohasida ($T < 100$ K) temperaturani 10 K qadam bilan oshirib borish orqali temperturaga bog'lanishi tadqiq etilgan.

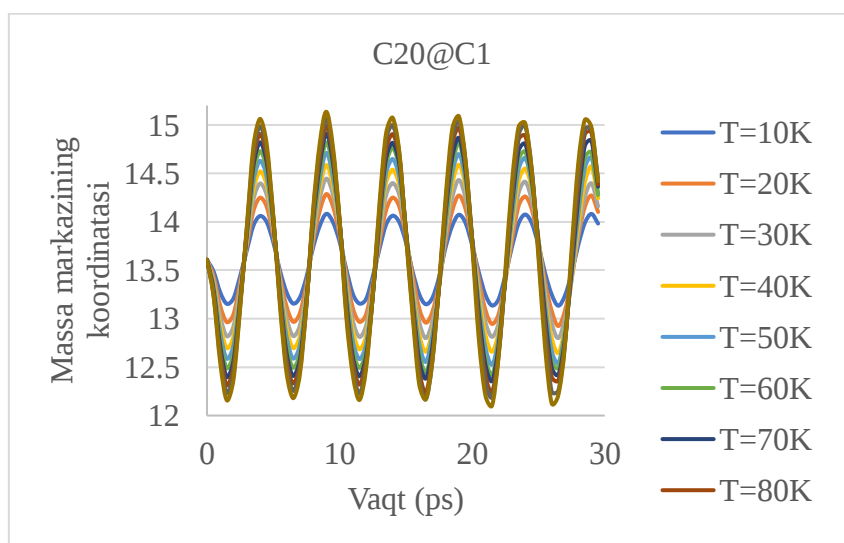
Yuqorida aytib o'tilgan modellashtirish ishlari LAMMPS ochiq paket dasturi [11] yordamida amalga oshirilgan. Olingan natijalarni vizualizatsiya qilish va ko'rib chiqalayotgan har bir harakat parametrlarini aniqlash uchun JMOL [12] kompyuter dasturidan foydalanilgan. C_{20} va C_{60} fulleren molekulalarining geometrik modellarining koordinatalari Nanotube Modeler [13] kompyuter dasturining ma'lumotlar bazasidan olingan. C_1 va C_3 uglerod klasterlarining C_{20} va C_{60} fulleren molekulalariga ekzodrillanishi, $C_{20}@C_1$, $C_{60}@C_1$ va $C_{20}@C_3$, $C_{60}@C_3$ (2-rasm) ekzofullerenlarining C_1 va C_3 uglerod klasterlari yordami Si(100)2×1 sirtida adsorbsiyasi LAMMPS paket dasturida MD usulida simulyatsiya qilindi. $34 \times 34 \times 15.2$ Å o'lchamli va 1083 atomdan iborat Si(100) taglik (1-rasm) dastlab LAMMPS paket dasturida yordamchi dastur yozish orqali olindi va undagi 2×1 rekonstruksiya adabiyotlarda [14] ta'kidlangan singari JMOL kompyuter dasturida olindi. Simulyatsiya qilinayotgan tizim temperaturasi, uning hajmi va undagi atomlar soni NVT ansambli yordamida boshqarildi.



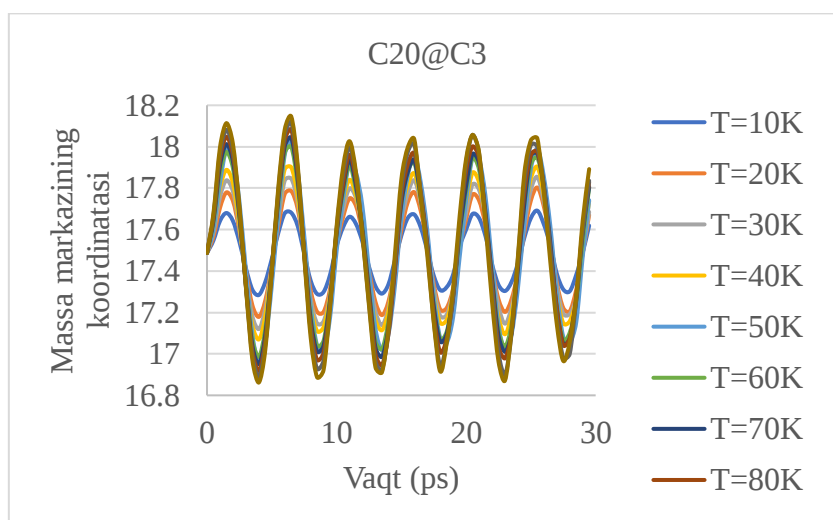
1- rasm. Si(100)2×1 taglikning yondan (a) va yuqoridan (b) ko'rinishi

Si(100)2×1 sirtining dimer qatorida C₂₀@C_n va C₆₀@C_n ekzofullerenlar C_n uglerod klasterlari bilan adsorbsiyalanib, C_n uglerod klasteridagi uglerod atomlari soni n=1, 3 va C_n klasterining C₂₀ va C₆₀ fulleren molekulari bilan ekzodrillanishi yagona C-C bog' orqali kuzatilganda fulleren molekulari Si(100)2×1 sirtida adsorbsiyalanmasdan, klasterning ekzodrillanishiga sabab bo'lgan C-C bog' bo'ylab o'tuvchi o'q atrofida aylanma va tebranma (presetsion) harakatlarni amalga oshiradi. Agar C_n uglerod klasteri undagi atomlar soni n=1, 3 bo'lib, fulleren molekulariga ikki va undan tadan ortiq sondagi C-C bog'lar yordamida ekzodrillansa, fulleren molekularining aylanma harakati kuzatilmagan, yagona tebranma harakati kuzatildi. Bunga sabab, fulleren molekularining aylanma harakatiga to'sqinlik qiluvchi potensial to'siq yakka bog'lanishli holatdagidan keskin ortib ketishidir. Agar uglerod klasterdagi atomlar soni n=2-5 va uning shakli chiziqli bo'lsa, fulleren molekulari kremniy taglik sirtiga yiqilib, fullerenli adsorbsiya kuzatilishi natijasida tadqiqotda o'rganilgan harakatlar kuzatilmadi. Shuningdek, C_n uglerod klasteridagi atomlar soni n=4, 5 bo'lib, klasterning shakli n=1, 3 bo'lgandagi kabi fulleren molekulasi yakkalik C-C bog' yordamida ekzodrillanishga imkon bergan hollarda, fulleren molekulari taglik sirtiga yiqilib, sirtida fullerenli adsorbsiyalanish ikkala harakatning ham sodir bo'lishiga imkon bermas ekan.

Tizim temperaturasi 10 K qadam bilan oshirib borilganda fulleren molekularining C-C bog' bo'ylab o'tuvchi AB o'q atrofida aylanma harakati jadallashib, ya'ni ushbu harakatning chastotasi ortib borishligi aniqlandi. Shuningdek, fulleren molekularining presetsion harakat chastotasining ham tizim temperaturasi bog'liqligi tizim temperaturasi 10 K qadam bilan oshirib borish orqali tadqiq qilinganda temperaturaning qaralayotgan sohasida ushbu harakat chastotasining o'zgarishsiz qolishi aniqlandi (2- va 3- rasmlar).

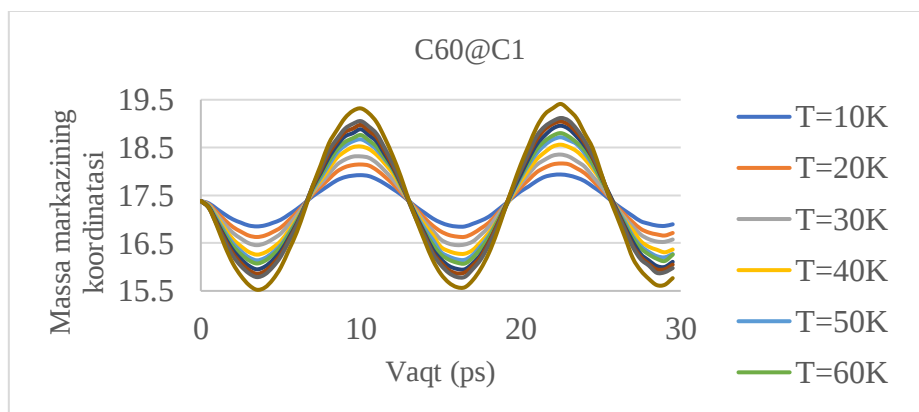


a)

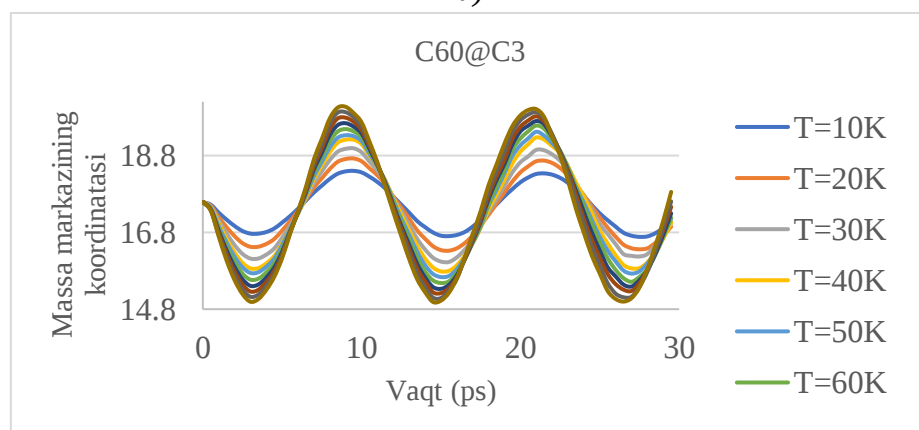


b)

2- rasm. $C_{20}@C_1$ va $C_{20}@C_3$ ekzofullerenlarining presetsion harakatadi C_{20} fulleren molekulasida massa markazi koordinatasining (x yoki y) vaqtga bog'lanishi.



a)



b)

3-rasm. $C_{60}@C_1$ va $C_{60}@C_3$ ekzofullerenlarining presetsion harakatadi C_{60} fulleren molekulasida massa markazi koordinatasining (x yoki y) vaqtga bog'lanishi.

2- va 3- rasmlardagi grafiklardan ko'rinib turibdiki, temperaturaning 10 K qadam bilan ortib borishi davomida $C_{20}@C_n$ va $C_{60}@C_n$ ekzofullerenlarining presetsion harakat chastotasi o'zgarishsiz

qolmoqda, faqat C_{20} fulleren molekulasi massa markazining tizim 0 K temperatura yaqinida bo'lganda egallaydigan holatdan og'ish masofasi (amplitudasi) ortib bormoqda. Mazkur og'ishlarning o'zgarish qiymatlari boshlang'ich temperatura o'zgarishlarida katta qiymatli va qaralayotgan temperatura intervalining oxiriga qarab esa kichik qiymatli bo'lib bormoqda. Bunga sabab sifatida fulleren molekulasining presetsion harakat kinetik energiyasining qaralayotgan og'ish qiymatlarining kvadratiga ($E \sim \Delta x^2$) proporsional ekanligini ko'rsatish mumkin.

Xulosa

Si(100) 2×1 sirtning dimer qatorida C_n uglerod klasterlari bilan adsorbsiyalangan $C_{20}@C_n$, $C_{60}@C_n$ ekzofullerenlarining $n=1, 3$ bo'lgan hollarda fullerenlar adsorbsiyasi mavjud bo'lmaganda C_{20} va C_{60} fulleren molekulalarining C_n uglerod klasterlarini fulleren molekulalariga ekzodrillovchi yakka C-C bog' bo'ylab o'tuvchi o'q atrofidagi aylanma harakat chastotasi past temperaturalar sohasida ($T < 100$ K) temperatura ortishi bilan ortib boradi, ammo ularning presetsion harakat chastotasi temperaturaga bog'liq emasligi kuzatildi.

Adabiyotlar

- [1]. Jing Li, Yang Cui and Lin Zhang, C_{60} adsorption on defective Si(100) surface having one missed dimer from atomic simulations at electrical level, *Arabian journal of chemistry*, Vol. 16, pp. 104816, 2023.
- [2]. M.L. Wang, X.Y. Sun, and X.Y. Hou, Role of buffer in organic solar cells using C_{60} as an acceptor, *Applied Physics letters*, Vol. 90, pp. 071109, 2007.
- [3]. I.Z. Urolov, F.F. Umarov, I.D. Yadgarov, G.T. Rakhmanov, Kh.I. Jabborov, Computer simulation study of adsorption processes of $C_{20}@C_n$ and $C_{60}@C_n$ ($n=1-5$) carbon clusters on reconstructed silicon Si(001) surface, *East European journal of physics*, Vol. 3, 273-280 pp., 2025.
- [4]. Kenneth S. Pitzer, Energy Levels and Thermodynamic Functions for Molecules with Internal Rotation, *The Journal of Chemical Physics*, Vol. 14, pp. 239-243, 1946.
- [5]. Hehre, W., Shusterman, G.; Shusterman, A. Conformational analysis of ethane, *Organic Chemistry*. Prentice Hall, 1996.
- [6]. Atkins, P., de Paula, J. Atkins, Describes the internal rotation barrier in ethane as a prototype torsional oscillator, *Physical Chemistry*. 10th ed., Oxford University Press, 2014.
- [7]. Eliel, E. L.; Wilen, S. H., Comprehensive discussion of conformations in propane, butane, pentane, etc., Butane's anti, gauche, and eclipsed conformations are covered in detail., *Stereochemistry of Organic Compounds*. Wiley, 1994.
- [8]. Fleming, I., Explains hyperconjugation and steric effects that control torsional preferences in alkanes, *Molecular Orbitals and Organic Chemical Reactions*. Wiley, 2009.

- [9]. Allinger, N. L., Conformational Analysis. 130. MM2., Journal of the American Chemical Society, Vol. 99, pp. 8127–8134, 1977.
- [10]. J. P. Lowe, The Barrier to Internal Rotation in Ethane, Science, Vol. 179, pp. 527-532, 1977.
- [11]. Sandia National Laboratories, Large-scale Atomic/Molecular Massively Parallel Simulator (LAMMPS), 2023, <https://www.lammps.org>
- [12]. Java, Jmol, 2023, <http://www.jmol.org>
- [13]. M.Yoshida, Nanotube Modeler (Nanocones, Bucky-Ball, Fullerenes, Simulation Software) (www.jcrystal.com)
- [14]. Ramstad, G. Brocks, and P. J. Kelly, Theoretical study of the Si(111) surface reconstruction, Physical review B, Vol. 51, pp. 14504-14523, 1995.